



Department of Physics
University of California San Diego

Modern Physics (2D)
Prof. V. Shrivastava
Quiz # 0 (Oct 7 2018)

Some Relevant Formulae, Constants and Identities

Edge of Observable Universe $\approx 10^{26} m$ away

$$\text{Centripetal Acc.} = \frac{u^2}{r}$$

Bohr's Angular Momentum Quantization: $mvr = n\hbar$

$$\text{Bohr Radius } a_0 = \frac{\hbar^2}{mke^2} = 0.529 \text{ \AA}$$

$$\text{Quantized Orbit in Hydrogenlike atom } r_n = \frac{n^2 a_0}{Z}$$

$$\text{Energy in Hydrogen atom } E_n = \frac{-ke^2}{2a_0} \left(\frac{1}{n^2} \right) = \left(\frac{-13.6 \text{ eV}}{n^2} \right)$$

$$\text{Gravitational Constant } G = 6.67 \times 10^{-11} \text{ N.m}^2/\text{Kg}^2$$

$$\text{Coulomb's Constant } k = 8.988 \times 10^9 \text{ N.m}^2/\text{C}^2$$

$$\text{Planck's Constant } h = 6.626 \times 10^{-34} \text{ J.s} = 4.136 \times 10^{-15} \text{ eV.s}$$

$$1 \text{ eV} = 1.60 \times 10^{-19} \text{ J}$$

$$\text{Electron Mass} = 9.1 \times 10^{-31} \text{ Kg} = 0.511 \text{ MeV}/c^2$$

$$\text{Proton Mass} = 938.3 \text{ MeV}/c^2$$

$$\text{Speed of Light in Vacuum } c = 2.998 \times 10^8 \text{ m/s}$$

$$\text{Electron Charge} = 1.602 \times 10^{-19} \text{ C}$$

Pl. write you answer in the Blue Book in indelible ink. Make sure your code number is prominently displayed on each page.



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Quiz # 4 (Feb 7)*

Problem 1 : Probing The Nucleus [6 pts]

Assume that Gold ($Z=79$) nucleus has a radius of 6.4 fm and the α -particle has radius of 1.8 fm. What minimum energy E must an incident α -particle have to experience non-Coulombic Nuclear forces, that is to say, penetrate the nucleus?

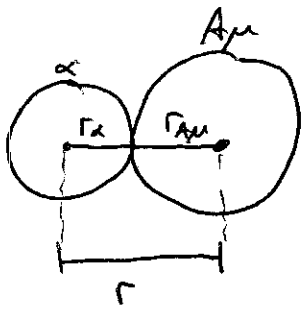
Problem 2: Basketball Team From Hell ! [14 pts]

Suppose an electron was bound to a proton, as in the hydrogen atom, by the gravitational force rather than by electric force. Write the equations for (a) the total energy of the system and (b) Newton's second law (equality of forces) . Now calculate (c) the radius and (d) the energy of the ground state. (e) Compare the size of the atom with that of our universe. Should humans made of such altered atoms be allowed in the basketball tournaments?

Phys 2D Quiz 4 Solns

1] When far away, $E = KE$ (since $PE = 0$ for large r).

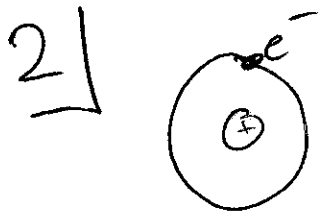
This E gets entirely converted to PE (for the min. energy case) when the nuclei touch.



$$\text{So } E = k \frac{\overset{\text{charge on } \alpha}{(2e)}(79e)}{r}$$

$$= k \frac{158e^2}{(r_\alpha + r_{A_\mu})}$$

$$= \boxed{4.43 \times 10^{-12} \text{ J} = 27.7 \text{ MeV}}$$



a]
$$E = \frac{1}{2} m_e v^2 - G \frac{M_e M_p}{r}$$

Since grav $PE = -G \frac{M_e M_p}{r}$

b] $F = ma \Rightarrow G \frac{M_e M_p}{r^2} = \frac{m v^2}{r}$

c] Assume $m_e v r = n \hbar$. So $v = \frac{n \hbar}{m_e r}$

Using part (b), we get

$$G \frac{m_e m_p}{r^2} = \frac{m_e}{r} \frac{n^2 \hbar^2}{m_e^2 r^2} \Rightarrow \boxed{r = \frac{n^2 \hbar^2}{G m_e^2 m_p}}$$

for $n=1$ (the ground state), $\boxed{r = 1.2 \times 10^{29} \text{ m}}$

d] Since by part (b) $G \frac{m_e m_p}{r} = m v^2$,

$$\boxed{E = -\frac{1}{2} G \frac{m_e m_p}{r}}$$

Using our answer ^{from} part (c) for r ,

$$\boxed{E = -4.2 \times 10^{-97} \text{ J} = -2.6 \times 10^{-78} \text{ eV}} \quad \text{DAMN.}$$

e] r is larger than the size of the universe (10^{26} m)!

\therefore It would be totally sweet if people made of these atoms played basketball.